

A Multiple-Quantum Nuclear Magnetic Resonance Investigation of Cure System in a Natural Rubber Compound

Jonathan Martens and Edward R. Terrill

Akron Rubber Development Laboratory, Inc.

Akron, OH



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Objective

- The purpose was the evaluation of cure system
 - Sulfur donor cure system vs. peroxide cure system
 - Better understand the cure mechanism
- Scope included Hahn Echo experiment and Multiple Quantum experiment

Outline

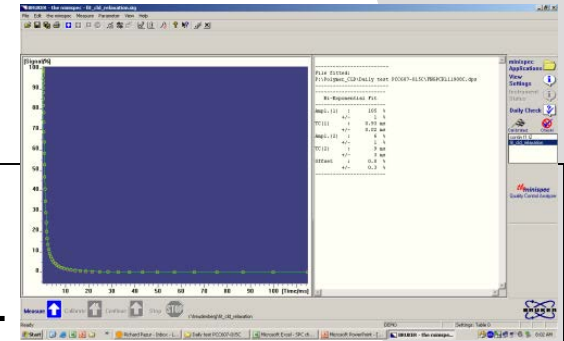
- Background on NMR
- Compounds
- Hahn Echo Results
- Double Quantum Results
- Conclusions

Background



minispec mq 20 NMR

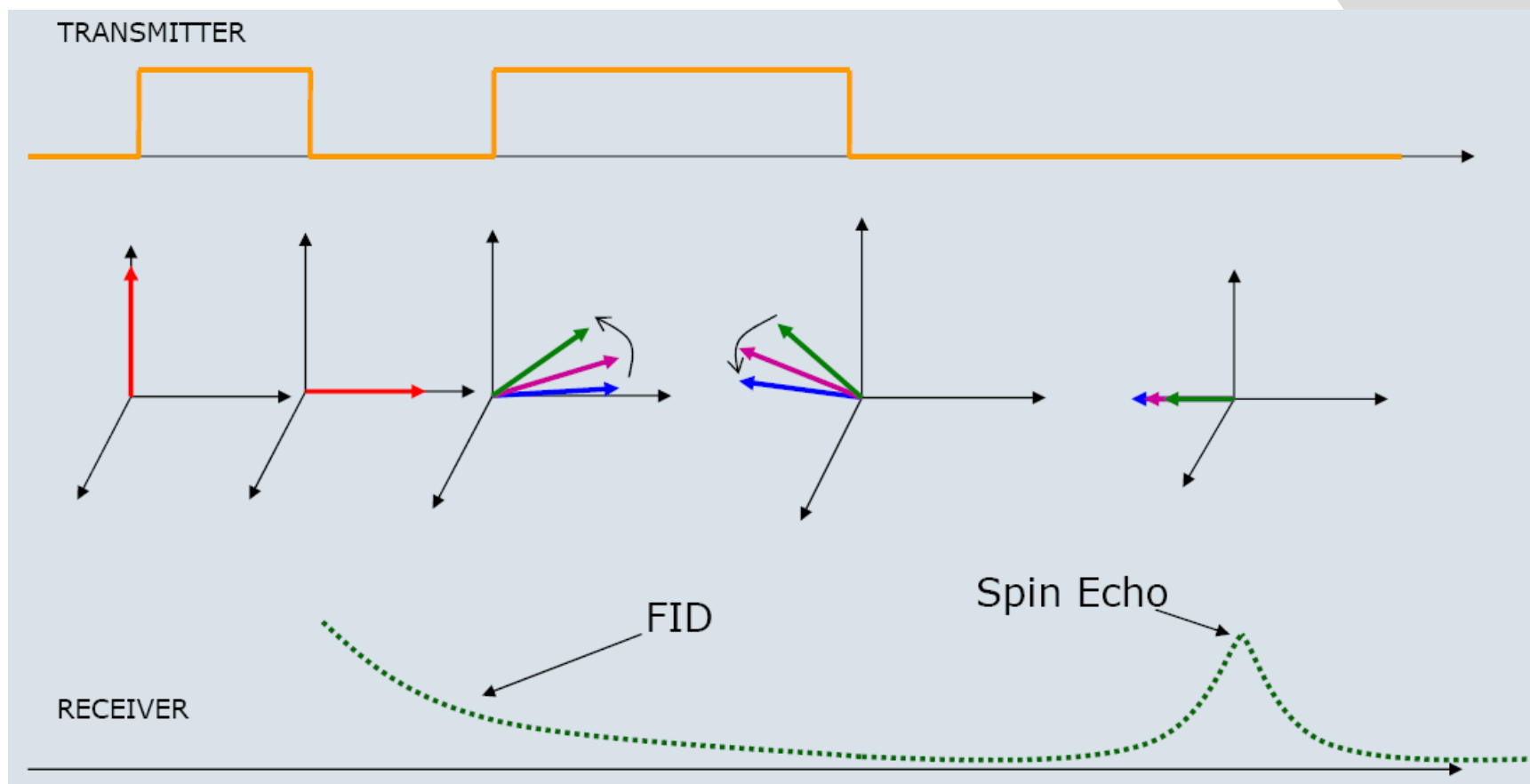
- 20 MHz for proton detection
- 0.5 Tesla



Advantages

- ✓ Low investment/maintenance cost
- ✓ Non-destructive technique
- ✓ Small amount of sample required ($\approx 250 \text{ mm}^3$)
- ✓ Quick measurement times (15- 20 min)

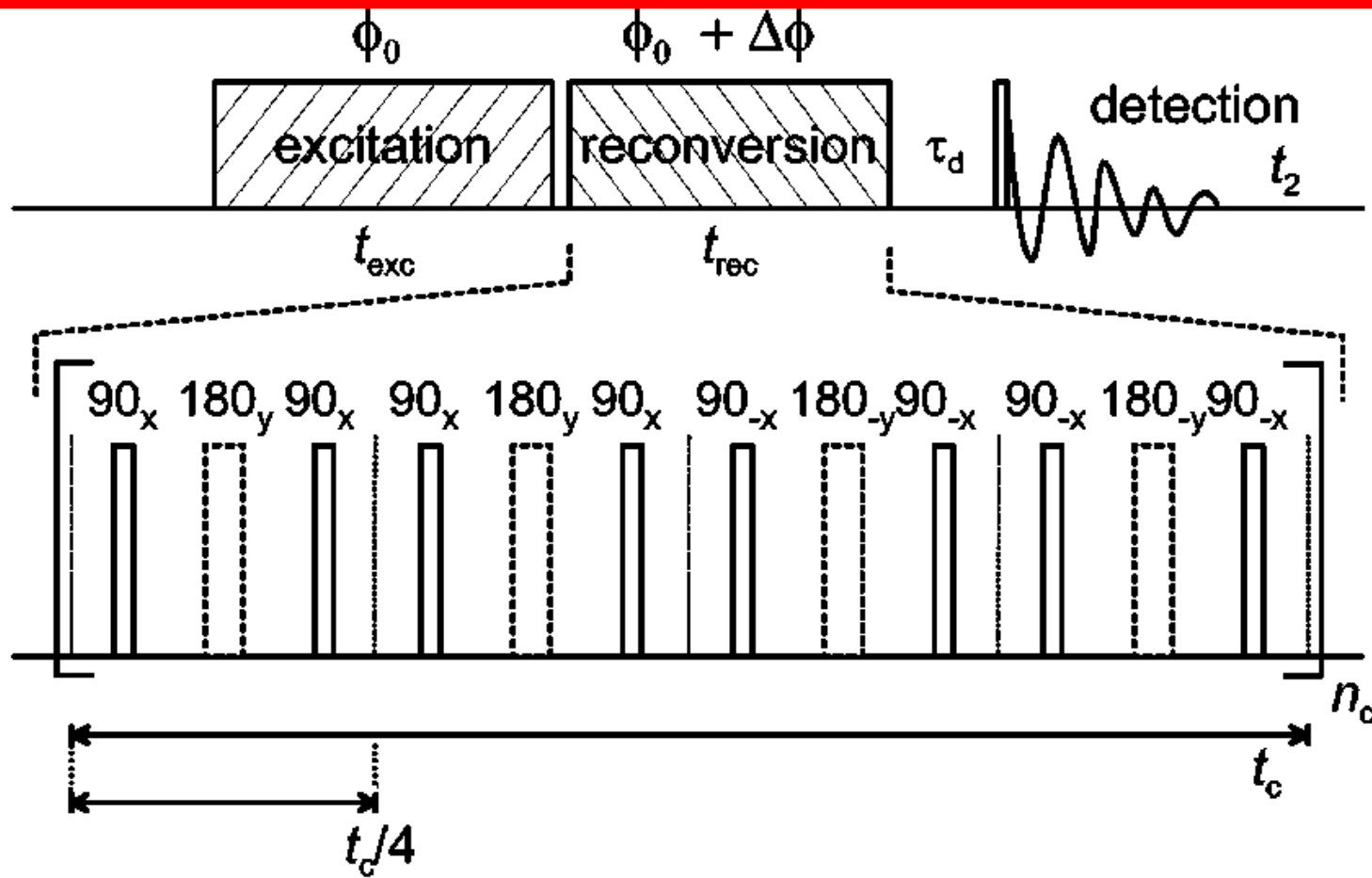
NMR Hahn Echo Spin Decay – T_2 measurement



Factors Affecting HE Spin-Spin Relaxation

1. Molecular entanglements – M_e
2. Chemical crosslinks – M_c
3. Fillers
4. Crystallinity
5. Plasticizers

Multiple Quantum Experiment



Details of Multiple Quantum

- Measures residual dipole couplings
- Relates to chain mobility (crosslink density)
- No effect of plasticizers

Compounds

Ingredient	P02	P04	P08	P12	P18	P24
NR	100					
ZnO	5					
Stearic Acid	2					
TMQ	1.5					
DCP (40% active)	0.5	1	2	3	4.5	6
Total PHR	109	109.5	110.5	111.5	113	114.5

Ingredient	SD1	SD2	SD3	SD4	SD5	SD6
NR	100					
ZnO	5					
Stearic Acid	2					
6PPD	1.5					
TMTD (75% active)	1.33	2.67	4	5.33	6.67	8
Total PHR	109.8	111.2	112.5	113.8	115.2	116.5

Hahn Echo Experiment

- Compounds: 100% Natural Rubber
- Special spin echo test for crosslink density

Biexponential Curve fitting

- $M(t) = A * \exp\left(\frac{t}{T_2} - q * M_2 * t^2\right) + B * \exp\left(\frac{t}{T_2}\right) + C$
 - $M_2 = 0.86 * 10^{10} \text{ s}^{-2}$ for Natural Rubber
 - Intermolecular dipole interactions to be determined from line shape analysis
 - A – relative amount of rigid component in rubber
 - B – relative amount of mobile component in rubber
 - C – curvefitting parameter
 - T_2 – represents decay time
 - q – correction factor for fast molecular motion



Crosslink Density Calculation

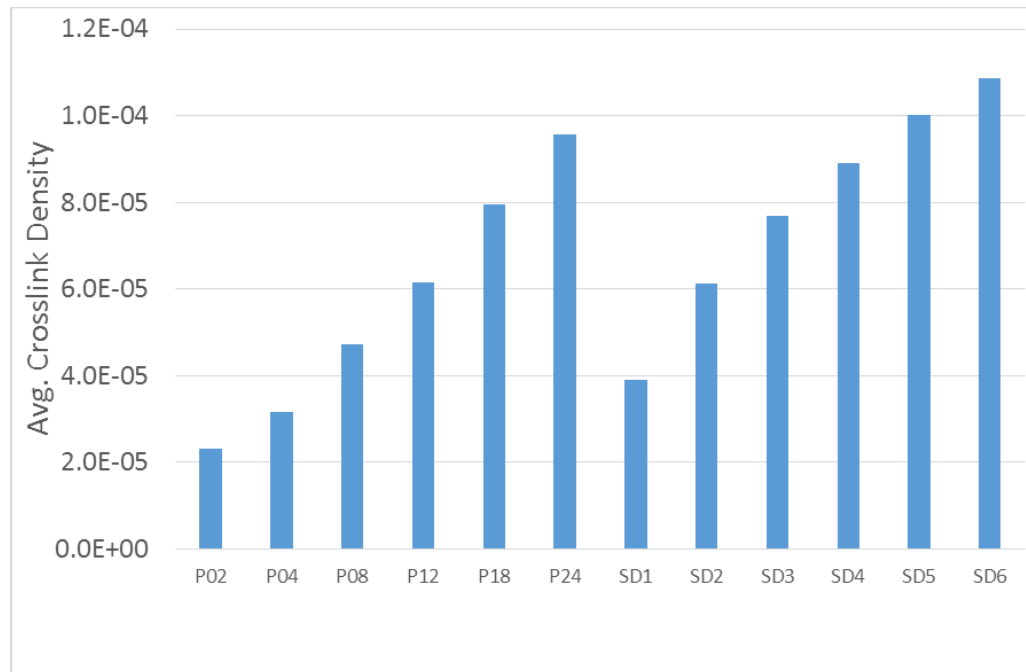
- $M_c = \frac{3}{5} * (q - q_0)^{-\frac{1}{2}} * c * \frac{M_{Ru}}{N}$
 - M_c – averaged inter-crosslink chain mass
 - q_0 – q-value for uncured sample
 - c – number of backbone units per Kuhn segment
 - $c = 7.4$ for Natural Rubber
 - M_{Ru} – molar mass of one monomeric unit
 - $M_{Ru} = 68$ g/mol for Natural Rubber
 - N – number of backbone bonds per monomeric unit
 - $N = 4$ for Natural Rubber

Crosslink Density Calculation

- $v_c = \frac{\rho}{M_c}$
 - v_c – chemical crosslink density
 - ρ – specific density of sample
 - M_c – average inter-crosslink chain mass

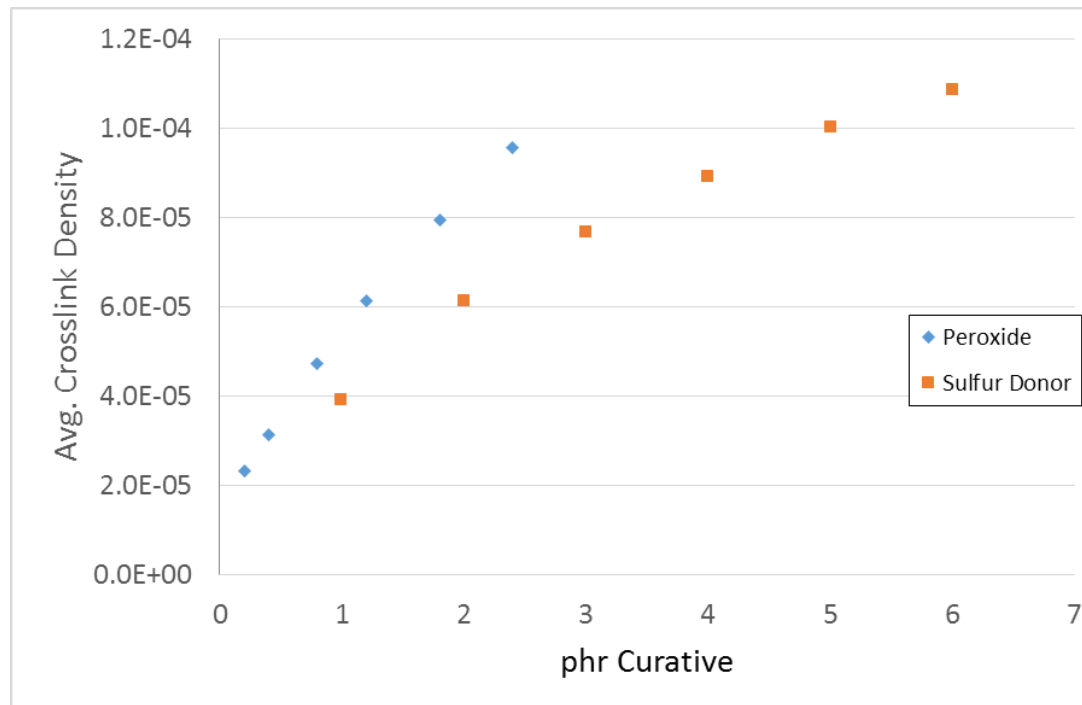
Crosslink Density Results

- Tested in triplicate
- Average of 3 tests shown



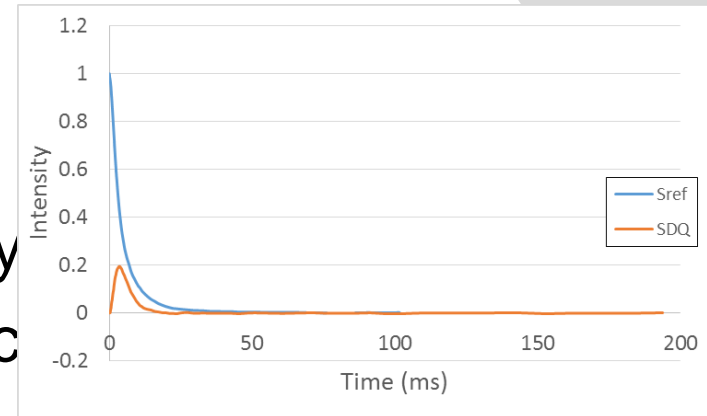
Crosslink Density Results

- Plot versus phr curative
- Compare similar loadings of curative



Double Quantum Experiment

- Two curve output
 - S_{DQ} and S_{ref}
 - S_{DQ} is DQ-filtered intensity
 - S_{ref} is reference intensity c



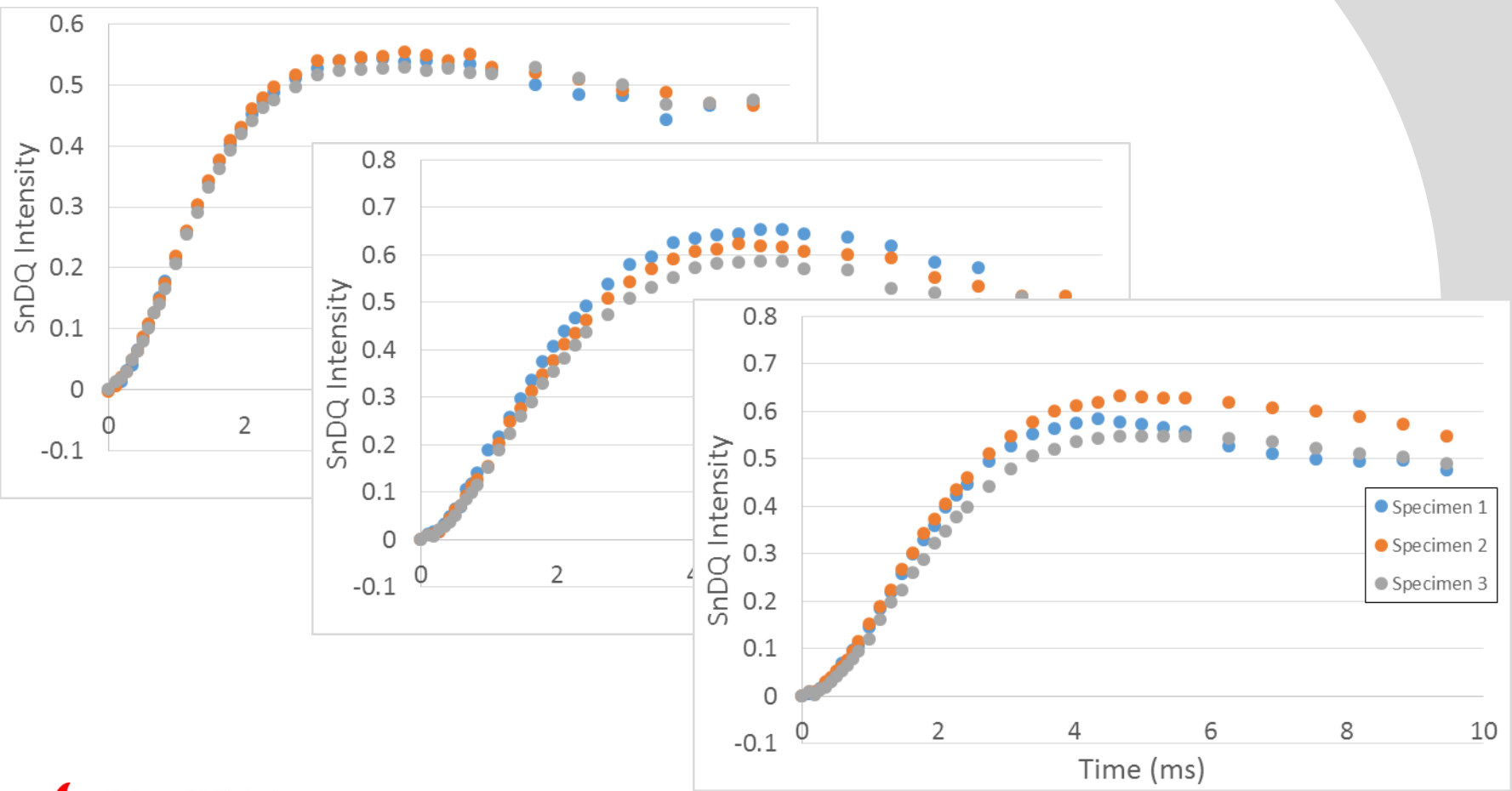
- $I_{DQ} = S_{DQ} + S_{ref}$

- Note: S_{ref} contains contributions from solvent, sol content, dangling chain ends, and other liquid-like content.

Removal of Contributions

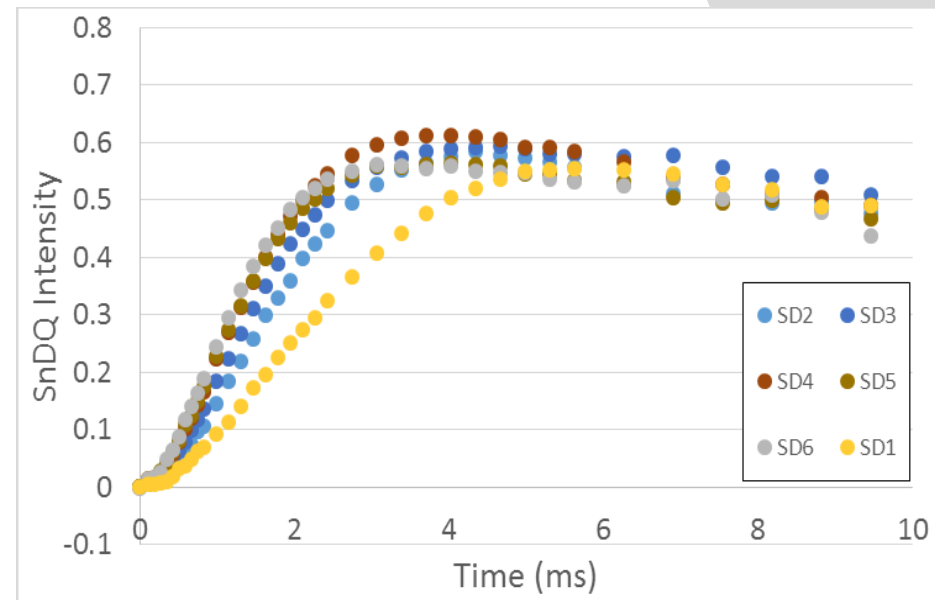
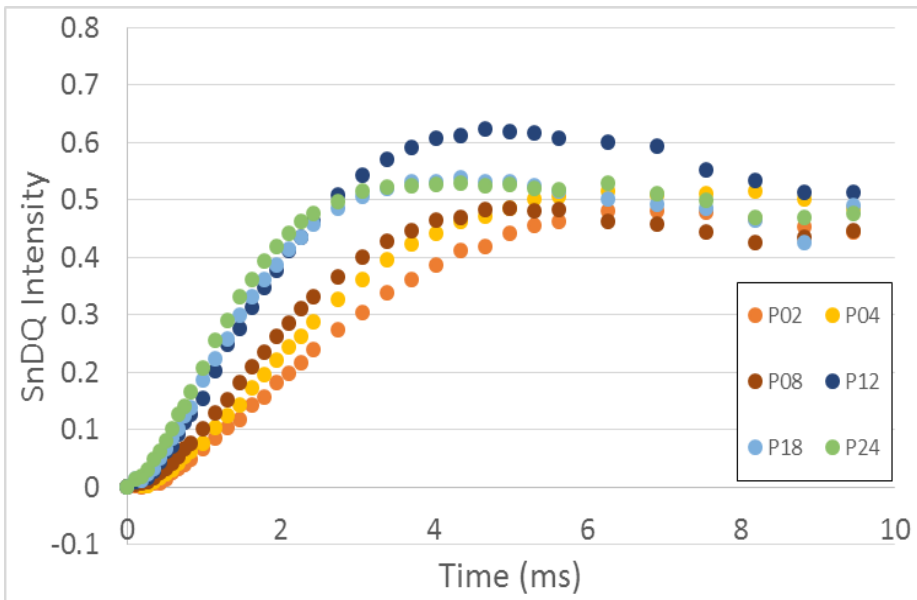
- Best fit end of S_{ref}
 - $C^* \exp(\tau_{\text{DQ}}/T_{2C})$
 - sol content
- $S_{\text{ref}} - S_{\text{DQ}} - C^* \exp(\tau_{\text{DQ}}/T_{2C})$
- Best fit end of residual curve
 - $B^* \exp(\tau_{\text{DQ}}/T_{2B})$
 - dangling chain ends
- $S_{\text{nDQ}} = S_{\text{DQ}} / (S_{\text{DQ}} + S_{\text{ref}} - B^* \exp(\tau_{\text{DQ}}/T_{2B}) - C^* \exp(\tau_{\text{DQ}}/T_{2C}))$

Normalized Build Up Curve



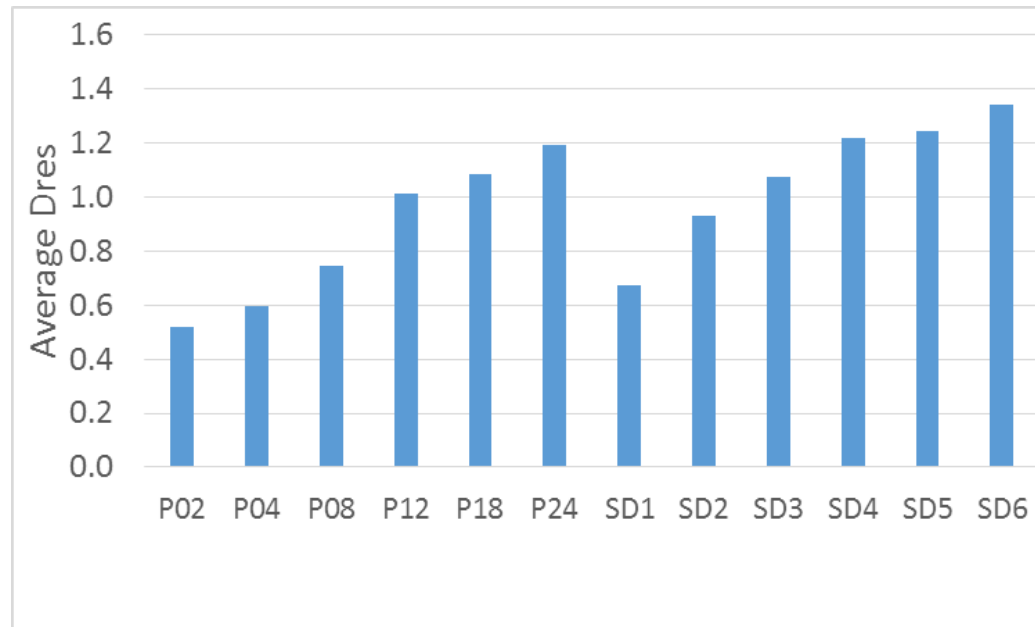
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Build Up Curve Comparison



Residual Dipole Coupling Constants (D_{res})

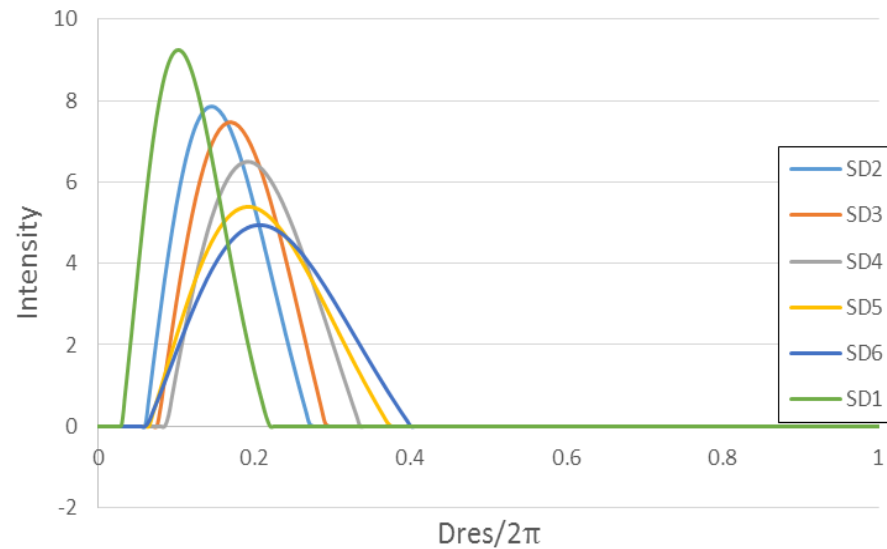
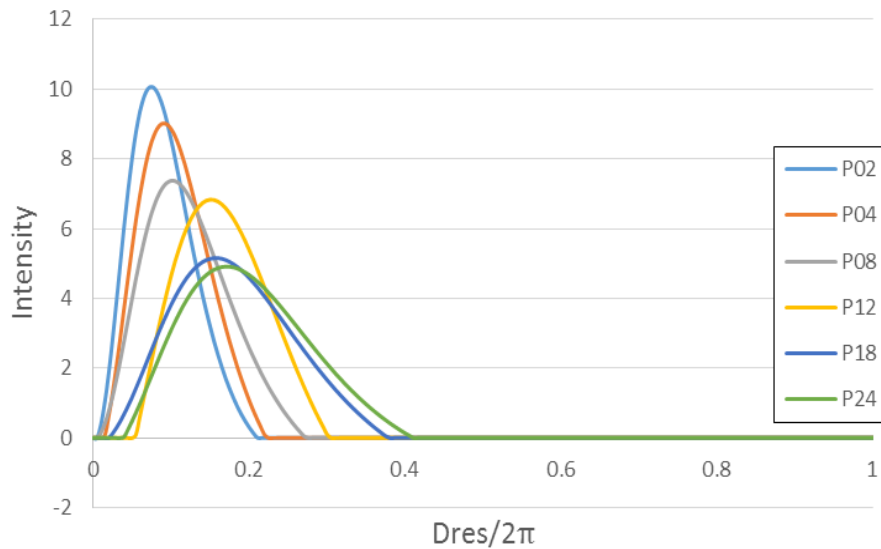
- Best fit of build up curve to 0.45 ms
- $$I_{nDQ}(\tau_{DQ}) = \frac{1}{2} \left(1 - \exp \left(-\frac{2}{5} * D_{res}^2 * \tau_{DQ}^2 \right) \right)$$



Molecular Weight Distribution of Crosslinks

- After Fast Tikhonov Regularization
- First 100 ms of build up curve
- Uses previous equation as Kernel Function

Molecular Weight Distribution of Crosslinks Comparison



Conclusions

- Effect of Loading:
 - Increasing crosslink density, (D_{res}) and broader distribution for both cure systems
- Effect of Cure System:
 - At corresponding phr loading, peroxide has higher crosslink density (D_{res}) than sulfur donor
 - At corresponding crosslink density, peroxide has broader distribution than sulfur donor

Acknowledgements

- Richard Pazur at QETE
 - Mixed compounds, offered advice, and validated our results
- Prof. Juan Valentin
 - Provided FTIKREG program and offered guidance and instruction on program and analysis of its output
- Prof. Kay Saalwächter
 - Provided DQ program



Questions?



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